

ph standards for calibration

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ph ph ph ph ph ph - ph

pH $\text{pH} = -\lg(c\text{H}^+)$ 1. ph ph ph ph ph

pH ph ph ph ph - ph

pH pH 0 14 7 25°C pH 7 pH ...

pH ph ph ph ph - ph

ph pH pH=7 c (H +) = 10⁻⁷ mol/L pH

ph pH ph ph ph ph ph - ph

pH 2 [Sørensen pH AH] pH 7.35~7.45

pH ph ph ph ph? pH 0 14 ph ph ph

14 pH 25 30 pH

ph PH ph ph ph ph - ph

ph PH pH 1909 ... 10

ph pH ph ph ph ph ph - ph

pH pH pH 6.5~8.5

ph pH 0 ph ph ph ph - ph

pH=0 pH pH=0 pH ...

ph Ph ph ph 0 - ph

ph ph pH= -lg c (H+) pw=10 14 c

$(H^+)=c(OH^-)=10^{-7}$ $p(H^+)=7$ $pH=7$

pH -

$pH = -\lg(c(H^+))$ $c(H^+) = 10^{-pH}$ $c(H^+) = 10^{-7} \text{ mol/L}$