

ph standards for calibration

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pH $\text{pH}=-\lg(\text{cH}^+)$ 1. ph ph ph ph ph

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pH $\text{pH}=-\lg(\text{cH}^+)$ 0 ph ph ph ph ph 14 ph ph ph ph ph 7 ph ph ph ph ph 25°C ph ph ph ph pH 7 ph ph ph pH ph ph ph ph pH ph ph ph ph ph ...

pH ph ph ph ph - ph

ph pH $\text{pH}=-\lg(\text{cH}^+)$ pH=7 c (H $^+$) = 10 $^{-7}$ mol/L ph ph ph ph pH

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ph pH $\text{pH}=-\lg(\text{cH}^+)$ 2 ph ph ph ph ph [Sørensen ph ph ph ph pH ph ph ph ph AH ph ph ph ph] 2 pH 7.35~7.45 ph ph

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$$[H^+]=[OH^-]=10^{-7}\text{mol/L}\quad p(H)=7\text{mol/L}\quad \text{pH}=7$$

pH -

$$pH = -\lg(c(H^+)) \quad c(H^+) = 1\text{ mol/L} \rightarrow pH = 0$$