

ph standards for calibration

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ph ph ph ph ph ph - ph

pH $\text{pH}=-\lg(\text{cH}^+)$ 1. ph ph ph ph ph

pH ph ph ph ph - ph

pH $\text{pH}=-\lg(\text{cH}^+)$ 0 ph ph ph ph ph 14 ph ph ph 7 ph ph ph ph
25°C ph ph ph pH 7 ph ph pH ph ph ph pH ph ph ph ...

pH ph ph ph ph - ph

ph pH $\text{pH}=-\lg(\text{cH}^+)$ pH=7 c
(H $^+$) = 10⁻⁷ mol/L ph ph ph pH

ph ph pH ph ph ph ph - ph

ph pH $\text{pH}=-\lg(\text{cH}^+)$ 2 ph ph ph ph ph [Sørensen ph ph ph pH
ph ph ph AH ph ph ph] 2 pH 7.35~7.45 ph ph

pH ph ph ph ph? pH ph 0 ph 14 ph ph ph

14 ph ph ph ph ph 14 pH ph ph pH 25 30 ph ph ph pH
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ph PH PH ph ph ph - ph

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ph ... ph ph 10 ph

ph pH ph ph ph ph - ph

ph pH $\text{pH}=-\lg(\text{cH}^+)$ pH ph ph pH ph pH
ph pH ph pH ph pH 6.5~8.5 ...

ph pH 0 ph ph ph ph - ph

ph pH=0 ph ph ph pH ph pH
ph pH=0 ph pH ph ph ph ...

ph Ph ph ph 0 ph - ph

ph ph ph pH $\text{pH}=-\lg(\text{cH}^+)$ pw=10 14 ph ph ph c

$$[H^+]=[OH^-]=10^{-7}\text{mol/L}\Rightarrow p(H)=7\text{mol/L}\Rightarrow p(H)=7$$

pH -

$$pH = -\lg(c(H^+)) \quad c(H^+) = 1\text{ mol/L} \Rightarrow pH = 0 \quad c(H^+) = 10^{-14}\text{ mol/L} \Rightarrow pH = 14$$