

is bleach a strong base

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Why is bleach a base? - Answers

No, bleach is not a strong acid. Bleach is a strong base, specifically a solution of sodium hypochlorite. It is commonly used as a disinfectant and for whitening purposes.

Is bleach an acid or base? - Answers

The conjugate base of a weak acid is generally a strong base, as in this case. Sodium hypochlorite is the main ingredient in bleach, a strong base.

Is household bleach an acid or a base? - Answers

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Sodium hypochlorite- acid or base? - Answers

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Which acid does bleach contain? - Answers

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Does bleach contain an acid - Answers

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What alkali does bleach contain? - Answers

Bleach contains Sodium hypochlorite and it is an alkali. Is bleach an alkali or an acid? Bleach is an alkaline substance. It has a high pH level, usually around 12-13, which makes it a strong base.

Is bleach a strong acid - Answers

No, bleach is not a strong acid. Bleach is a strong base, specifically a solution of sodium hypochlorite. It is commonly used as a disinfectant and for whitening purposes. Anonymous • Lvl 1 •

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Is bleach a acid or alkali? - Answers

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Is sodium hypochlorite an acid or alkaline? - Answers

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